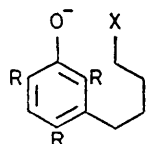


## Regiospecific Alkylation of Phenols: *ortho* or *para* to $\alpha$ Coupling in Cyclization of Bis-phenols via Quinone Methides

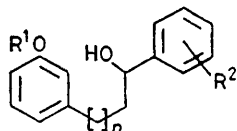
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**Summary** Intramolecular alkylations of phenols by *o*- and *p*-hydroxybenzylic alcohols are achieved with high regioselectivity by heating the bis-magnesium salts of bis-phenols such as (2) and (5), which cyclise efficiently into aryl-indanols and aryltetralols.

FOLLOWING our successful  $\text{Ar}_2^-6$  cyclisation of phenoxide anions of general type (1) using achiral<sup>1</sup> and chiral<sup>2</sup> leaving groups, we attempted the analogous cyclisation involving an enone electron sink. Quinone methides were considered likely candidates since (a) the precursors are readily prepared, (b) aromatization provides a strong driving force, and (c) studies<sup>3</sup> on intermolecular systems indicated that intramolecular magnesium(II) chelation could induce regioselectivity.



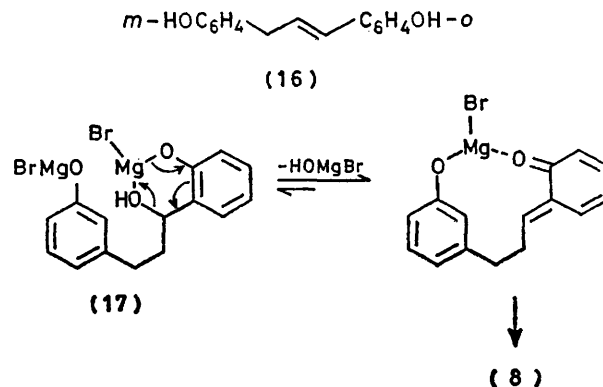
(1)  
a; R = H  
b; R = Me



(2)  $n = 1, R^1 = H, R^2 = o\text{-OH}$   
(3)  $n = 1, R^1 = H, R^2 = o\text{-OMe}$   
(4)  $n = 1, R^1 = \text{Me}, R^2 = o\text{-OH}$   
(5)  $n = 2, R^1 = H, R^2 = o\text{-OH}$   
(6)  $n = 1, R^1 = H, R^2 = p\text{-OH}$   
(7)  $n = 2, R^1 = H, R^2 = p\text{-OH}$

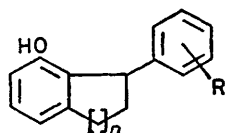
The carbinol (2)† (1 mol) was heated under reflux for 20 h in benzene with ethylmagnesium bromide<sup>3,4</sup> (2 mol). The bis-phenol (8)† was formed (71%)‡ together with the dehydration product (16) (11%).‡ G.l.c.§ of the dimethyl ether of (8) indicated that only a trace (< 1%) of (12) had been formed. In contrast, when the carbinol (2) was cyclised ( $\text{SnCl}_4, \text{CH}_2\text{Cl}_2$ ), a mixture (72%)‡ of (8) and (12) in the ratio 5:95 respectively was formed. The striking regioselectivity of the  $\text{Mg}^{\text{II}}$ -induced cyclisation is accounted for in the mechanism in the Scheme. The evidence for this is based on the following results: (a) (2) did not cyclise when heated with either NaH or BuLi in benzene, (b) the bis-magnesium salt (17) (1 mol) did not cyclise under

standard conditions when 18-crown-6 (2 mol) was present, (c) (16) (1 mol) was recovered unchanged when heated with ethylmagnesium bromide (2 mol) in benzene, (d) under standard conditions neither (3),† (4),† nor (18)† cyclised, and (e) the *o*-quinone methide formed from the phenol (19) was trapped with ethyl vinyl ether<sup>5</sup> to yield the cycloadduct (20) (55%).‡ High *ortho*-regioselectivity is again evident in the cyclisation of the homologue (5).† A mixture of (9)† and (13)† in the ratio 88:12§ was isolated (90%).‡ The increased percentage of *para*-alkylation indicates a more weakly complexed quinone methide intermediate than that derived from (17) (Scheme). The ease of six-membered ring formation is highlighted by the high yield (90%)‡ of cyclised product obtained from (5) (1 mol) even in the presence of 18-crown-6 (2 mol). In this instance the ratio of (9) to (13) was 38:62.§

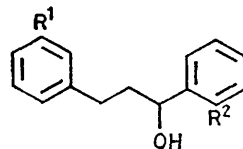
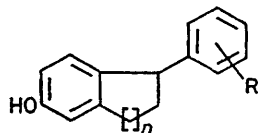


SCHEME

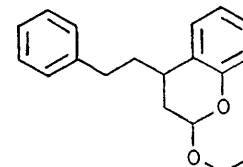
The triols (6)† and (7)† were investigated to probe the characteristics of *p*-quinone methide cyclisations. Since intramolecular  $\text{Mg}^{\text{II}}$  bridging would not then be involved, we expected (a) inefficient cyclisation of (6), (b) efficient cyclisation of (7), and (c) higher *para*- to *ortho*-regioselectivity. In the event, (6) cyclised (43%)‡ to (10)† and (14)† in the ratio 10:90§ and (7) cyclised (90%) to (11)† and



(8)  $n = 1, R = o\text{-OH}$  (12)  
(9)  $n = 2, R = o\text{-PH}$  (13)  
(10)  $n = 1, R = p\text{-OH}$  (14)  
(11)  $n = 2, R = p\text{-OH}$  (15)



(18)  $R^1 = \text{OH}, R^2 = \text{H}$   
(19)  $R^1 = \text{H}, R^2 = \text{OH}$



(20)

† The <sup>1</sup>H n.m.r. and analytical data are in accord with this structure.

‡ Isolated yield.

§ Determined by g.l.c. of the dimethyl ethers using a 2 m column of 2½% CEMS on Chromosorb G with a programmed temperature increase in the range 150–200°C.

(15)† in the ratio 7:93.§ By comparison, cyclisation (SnCl<sub>4</sub>, CH<sub>2</sub>Cl<sub>2</sub>) of the dimethyl ether of (7)† gave a product mixture (71%)‡ with an *ortho* to *para* ratio of 10:90.§ The high *para*-regiospecificity of *p*-quinone methides is most probably due to steric hindrance of the *ortho*-position by the magnesium cation in the bis-Mg<sup>II</sup> phenoxide salts of both

(6) and (7). Thus, for example, cyclisation of the bis-Mg<sup>II</sup> salt of (7) (1 mol) in the presence of 18-crown-6 (2 mol) occurred (83%)‡ with the formation of (11) and (15) in the ratio of 40:60§ respectively.

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<sup>1</sup> W. S. Murphy and P. G. Duggan, *J.C.S. Perkin II*, 1975, 1054.

<sup>2</sup> P. G. Duggan and W. S. Murphy, *J.C.S. Perkin I*, 1976, 634.

<sup>3</sup> G. Casiraghi, G. Casnati, M. Cornia, A. Pochini, G. Sartori, and R. Ungaro, *J.C.S. Perkin I*, 1978, 322.

<sup>4</sup> J. A. Asgill, L. Crombie, and D. A. Whiting, *J.C.S. Chem. Comm.*, 1978, 59.

<sup>5</sup> D. A. Bolon, *J. Org. Chem.*, 1970, 35, 3666.